



Electromagnetics:
Electromagnetic Field Theory

Forces on Point Charges



1

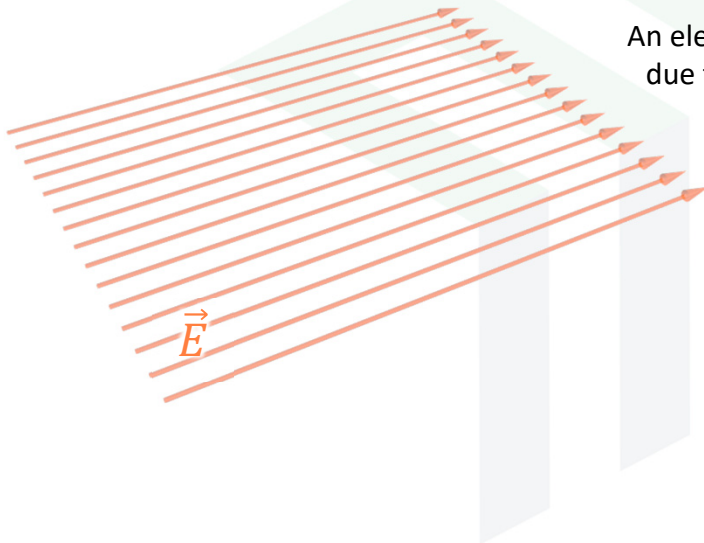


Force on a Point Charge

Side 2

2

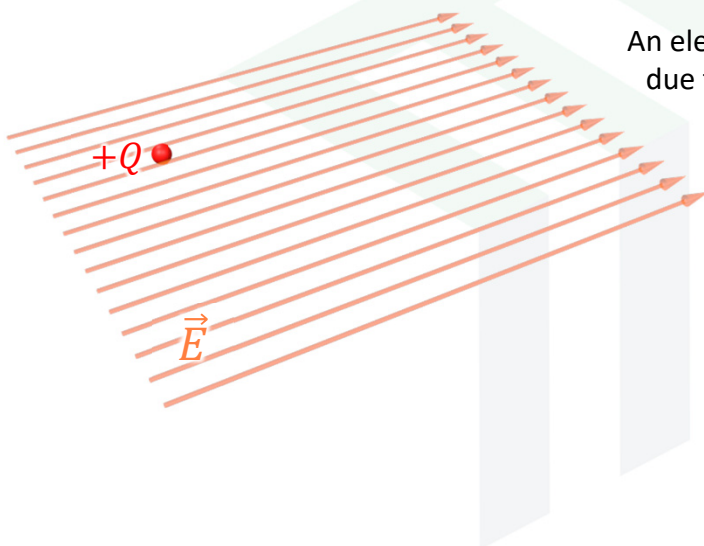
Force \vec{F} on a Point Charge Q



An electric field \vec{E} may be present due to other charges, an applied voltage, a wave, etc.

3

Force \vec{F} on a Point Charge Q

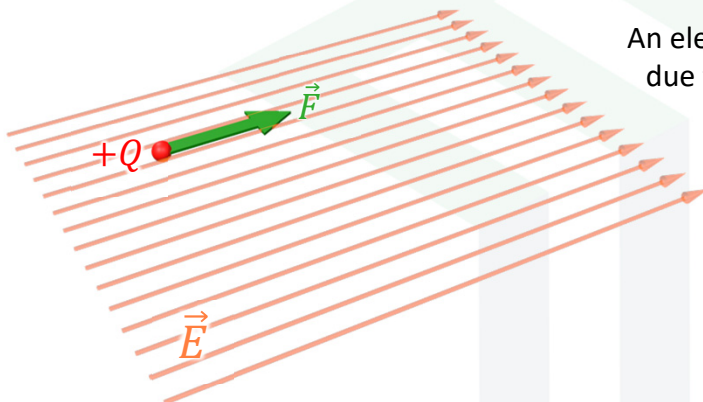


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Let there be a charge Q .

4

Force \vec{F} on a Point Charge Q




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Let there be a charge Q .

$$\vec{F} = Q\vec{E}$$

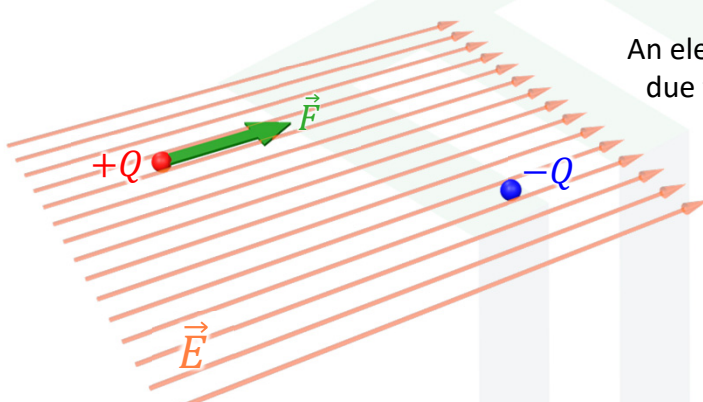
$\vec{F} \equiv$ force vector (N)

Force is most closely associated with electric field intensity \vec{E} .

 Slide 5

5

Force \vec{F} on a Point Charge Q




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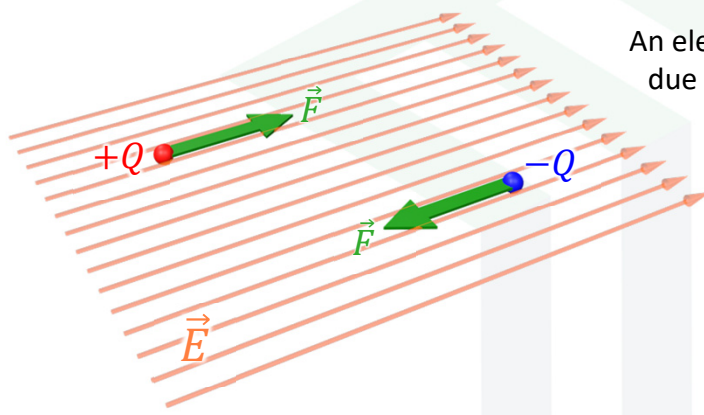
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 Slide 6

6

Force \vec{F} on a Point Charge Q



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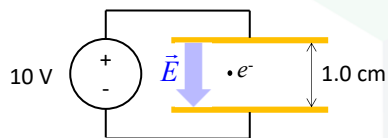
7

Example

What is the force \vec{F} on an electron located between two metal plates separated by 1 cm and 10 volts is applied?

Solution

Sketch the problem.



Calculate the electric field intensity \vec{E} .

$$\vec{E} = -\frac{V}{d}\hat{a}_z = -\frac{10\text{ V}}{0.01\text{ m}}\hat{a}_z = -1000\hat{a}_z\text{ V/m}$$

Calculate the force on the electron.

$$\begin{aligned}\vec{F} &= Q\vec{E} \\ &= (-1.60 \times 10^{-19}\text{ C})(-1000\hat{a}_z\text{ V/m}) \\ &= 1.60 \times 10^{-16}\hat{a}_z\text{ N} \\ &= 160\hat{a}_z\text{ aN}\end{aligned}$$

Attonewtons (wow, that is small!!)

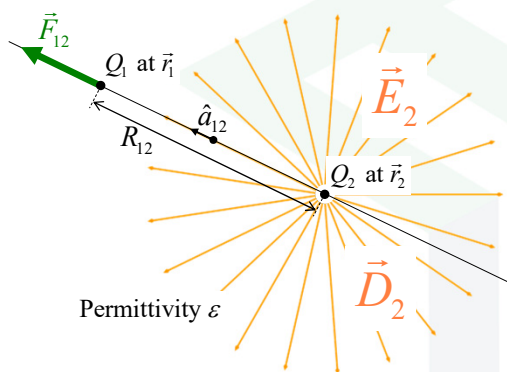
8

Coulomb's Law

Slide 9

9

Coulomb's Law



Force on Q_1 due to Q_2 is

$$\vec{F}_{21} = \frac{Q_1 Q_2}{4\pi\epsilon R_{12}^2} \hat{a}_{12} = \frac{Q_1 Q_2}{4\pi\epsilon} \frac{\vec{r}_2 - \vec{r}_1}{|\vec{r}_2 - \vec{r}_1|^3}$$

Given two charges, Q_1 and Q_2 .

Charge Q_2 creates an electric flux \vec{D}_2 .

$$\vec{D}_2 = \frac{Q_2 (\vec{r} - \vec{r}_{Q_2})}{4\pi |\vec{r} - \vec{r}_{Q_2}|^3}$$

The electric field intensity \vec{E}_2 is calculated from the electric flux density \vec{D}_2 .

$$\vec{E}_2 = \frac{1}{\epsilon} \vec{D}_2$$

Electric field \vec{E}_2 puts a force on charge Q_1 .

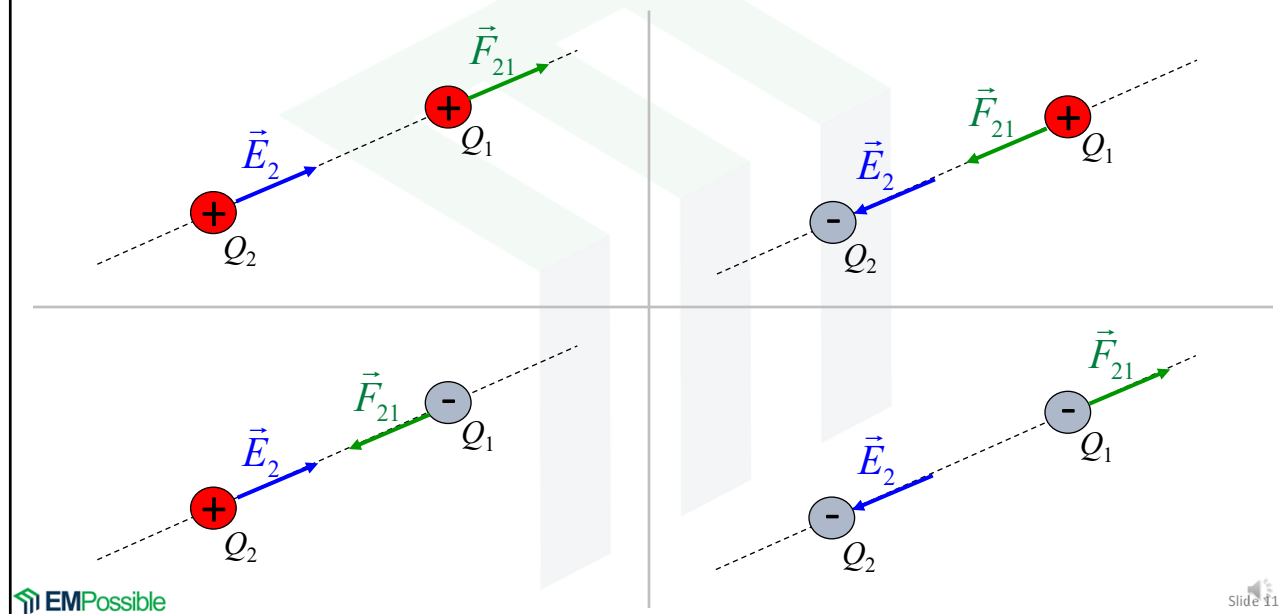
$$\vec{F}_{21} = Q_1 \vec{E}_2$$

EMPossible

Slide 10

10

Summary of Direction of Force on Q_1 Due to Q_2



11

Notes About Coulomb's Law

- Direction of force is along the line joining the two charges.
- Force is directly proportional to $Q_1 Q_2$.
- Force is inversely proportional to R^2 .
- Charges must be stationary for these equations to be valid (i.e. electrostatics).

EMPossible

Slide 12

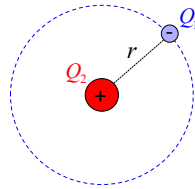
12

Example

Calculate the attractive force between the electron and proton in a hydrogen atom.

Solution

Sketch the problem.



Charge:
 $Q_1 = -1.60 \times 10^{-19} \text{ C}$
 $Q_2 = +1.60 \times 10^{-19} \text{ C}$
 Distance:
 $r \approx 53 \text{ pm}$

Calculate the force

$$\begin{aligned}\vec{F}_{21} &= \frac{Q_1 Q_2}{4\pi\epsilon_0\epsilon_r R_{12}^2} \hat{a}_{12} \\ &= \frac{(-1.60 \times 10^{-19} \text{ C})(+1.60 \times 10^{-19} \text{ C})}{4\pi(8.854 \times 10^{-12} \text{ F/m})(1.0)(53 \times 10^{-12} \text{ m})^2} \hat{a}_r \\ &= -8.19 \times 10^{-8} \hat{a}_r \text{ N} \\ &= \boxed{82 \text{ nN}}\end{aligned}$$

What About Time t ?

At $t = 0$, a charge Q with mass m is located at the origin and moving with an initial velocity of v_0 in the $+x$ direction. At this instant in time, a uniform electric field $\vec{E} = E_0 \hat{a}_y$ is applied. Derive an expression for the position of the charge as a function of time for $t \geq 0$.

Force due to electric field. $\vec{F}(t) = Q(t)\vec{E}(t)$

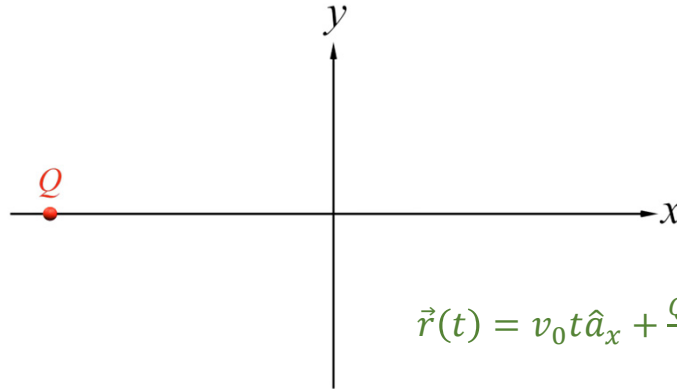
Acceleration $\vec{F}(t) = m(t)a(t) \quad a(t) = \frac{\vec{F}(t)}{m(t)} = \frac{Q\vec{E}}{m} = \frac{QE_0\hat{a}_y}{m}$

Velocity $\vec{v}(t) = \vec{v}_0 + \int_{-\infty}^t a(\tau) d\tau = v_0 \hat{a}_x + \int_0^t \frac{QE_0 \hat{a}_y}{m} d\tau = v_0 \hat{a}_x + \frac{QE_0 t \hat{a}_y}{m}$

Position $\vec{r}(t) = \vec{r}_0 + \int_{-\infty}^t v(\tau) d\tau = \int_0^t \left(v_0 \hat{a}_x + \frac{QE_0 \tau \hat{a}_y}{m} \right) d\tau = v_0 t \hat{a}_x + \frac{QE_0 t^2}{2m} \hat{a}_y$

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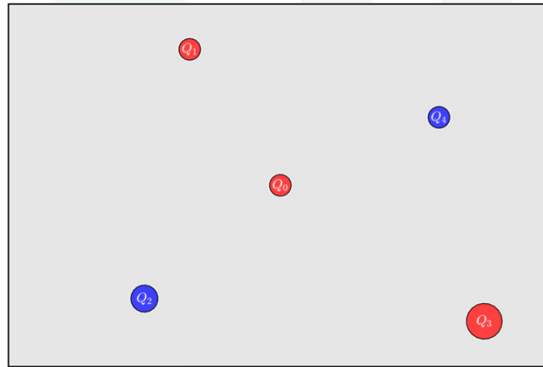


Coulomb's Law for Multiple Point Charges

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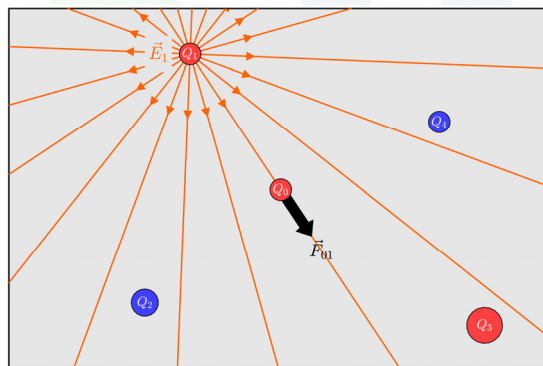
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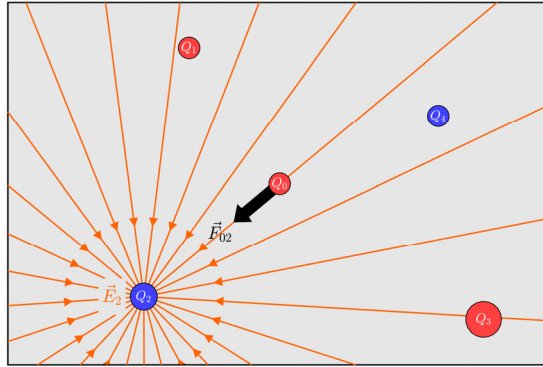
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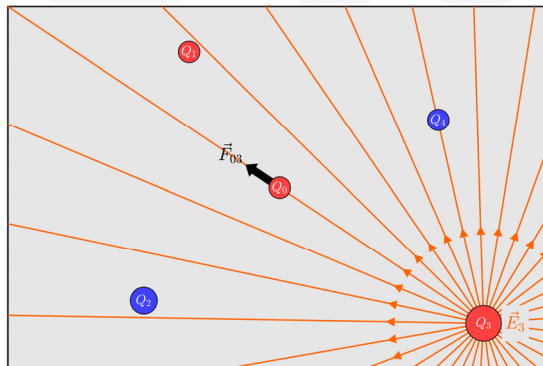
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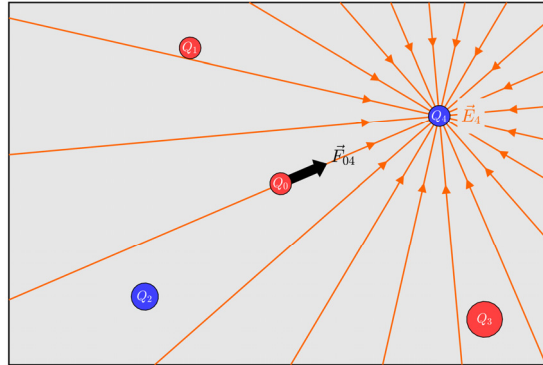
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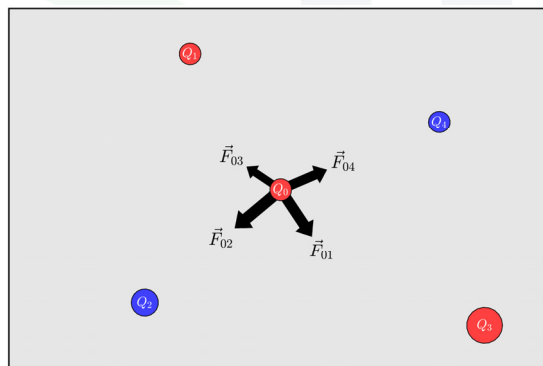


21

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22

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